

# FORMATION OF SECONDARY PM IN CAPITAL REGION

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# OUTLINE

- Background (Capital Region, monitoring data)
- Modeling Methodology
- Results
- Source attribution
- Conclusion

# BACKGROUND: PARTICULATE MATTER (PM)

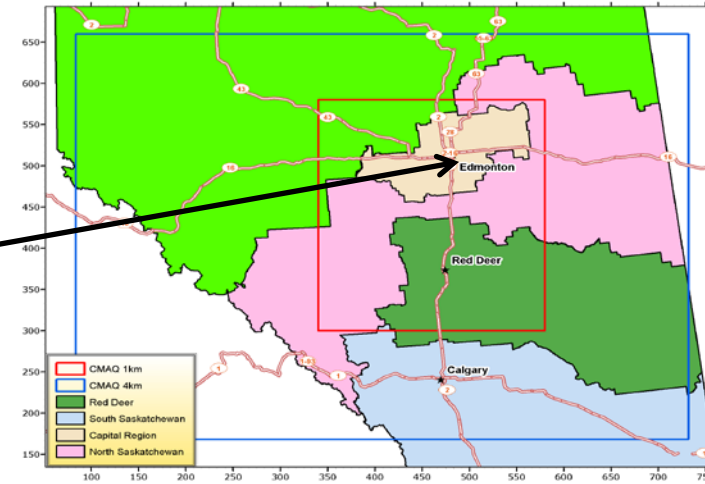
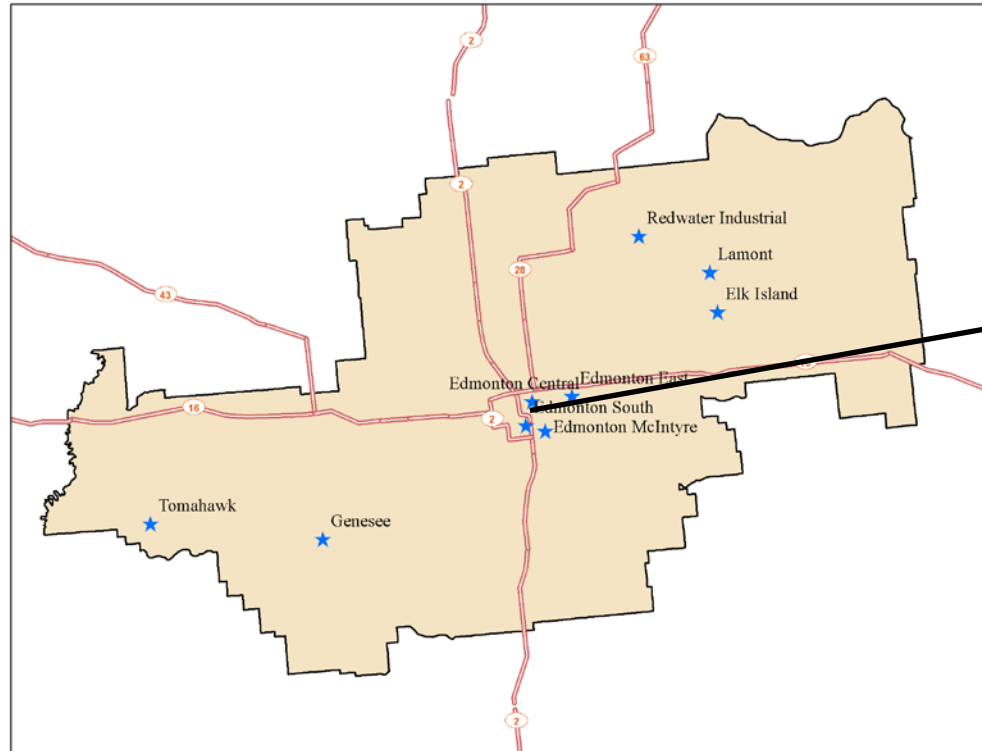
- Fine PM : diameter smaller than 2.5  $\mu\text{m}$ ; complex mixture consisting of *many* different *components*
- Cause health, vegetation, visibility problems
- Canadian Ambient Air Quality Standards (CAAQS)

Pollutants	Old Standards	New Standards	
		2015	2020
PM <sub>2.5</sub> Annual	-	10 $\mu\text{g}/\text{m}^3$	8.8 $\mu\text{g}/\text{m}^3$
PM <sub>2.5</sub> 24-hour	30 $\mu\text{g}/\text{m}^3$	28 $\mu\text{g}/\text{m}^3$	27 $\mu\text{g}/\text{m}^3$

- Origins
  - Primary: emitted from a source
  - **Secondary**: formed through chemical and physical reactions involving different precursor gases



# BACKGROUND: CAPITAL REGION

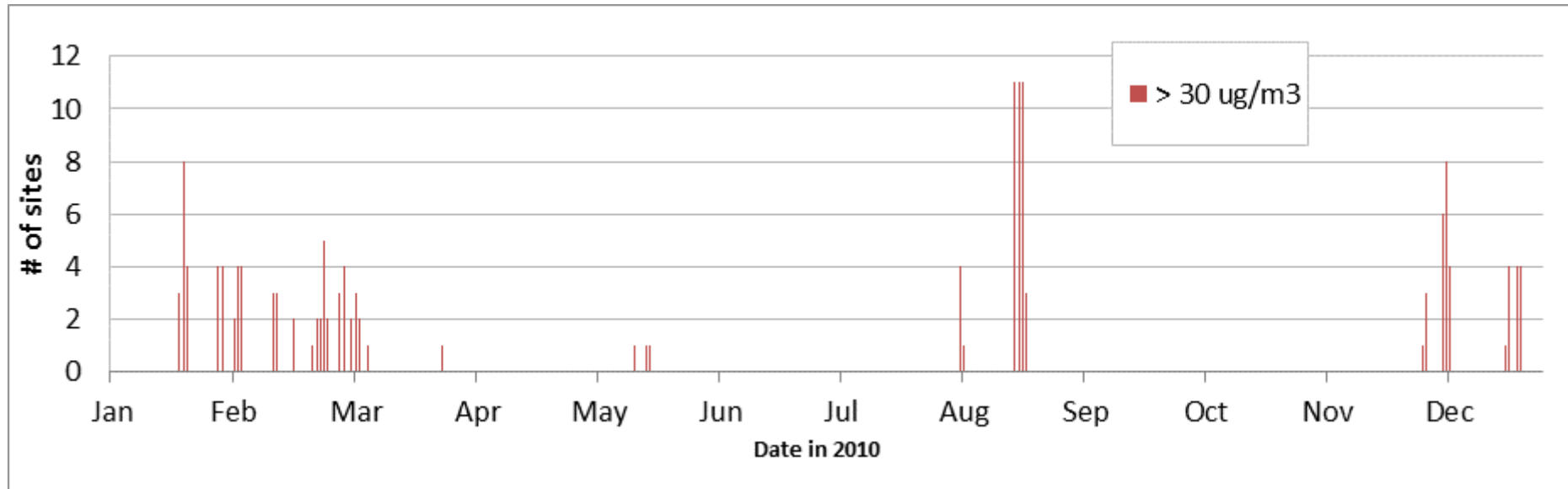


Sources in Capital Region:  
Mobile, EGU, Industrial sources, Agriculture

24-hr average  $> 30 \mu\text{g}/\text{m}^3$

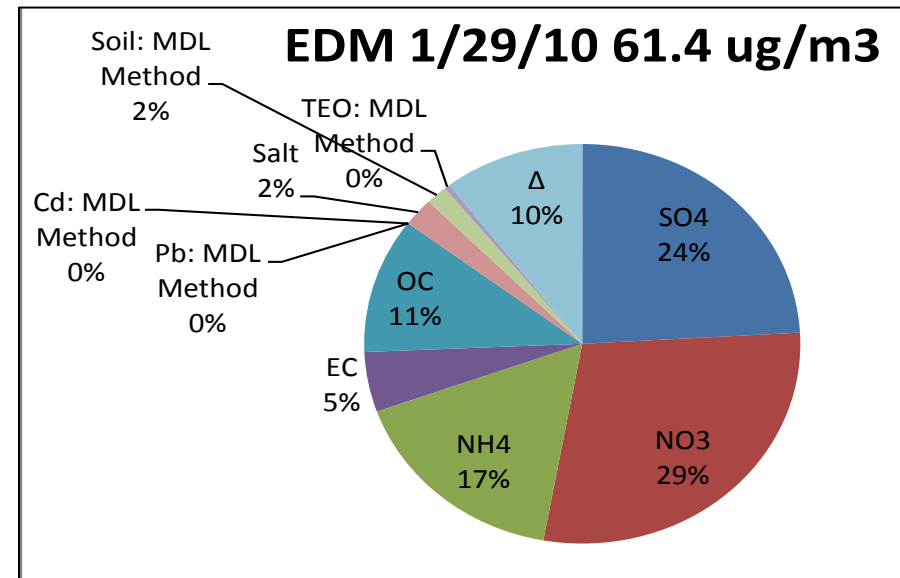
- 2008 : 9 exceedance days; all in winter
- 2009 : 7 exceedance days; 5 in winter
- 2010 : **41** exceedance days; 4 due to fires

# BACKGROUND: 2010 EXCEEDANCES



- 9 winter episodes : peak on Jan 29
- Winter episodes characterized by higher than typical secondary PM<sub>2.5</sub>

Non-event :  $\text{NH}_4\text{SO}_4 + \text{NH}_4\text{NO}_3 \sim 30\%$   
 24-hr PM<sub>2.5</sub> < 10  $\mu\text{g}/\text{m}^3$



# BACKGROUND: SUMMER VS WINTER PM

## Summer

Active Photochemistry

Strong vertical mixing

Dominated by secondary PM

OM > SO<sub>4</sub> > NO<sub>3</sub>

High PM typically associated with SO<sub>4</sub>

## Winter

Slow photochemistry (scarce sunlight and low radical availability)

Light winds, temperature inversion → accumulation of pollution

Dominated by secondary PM

OM > NO<sub>3</sub> > SO<sub>4</sub>

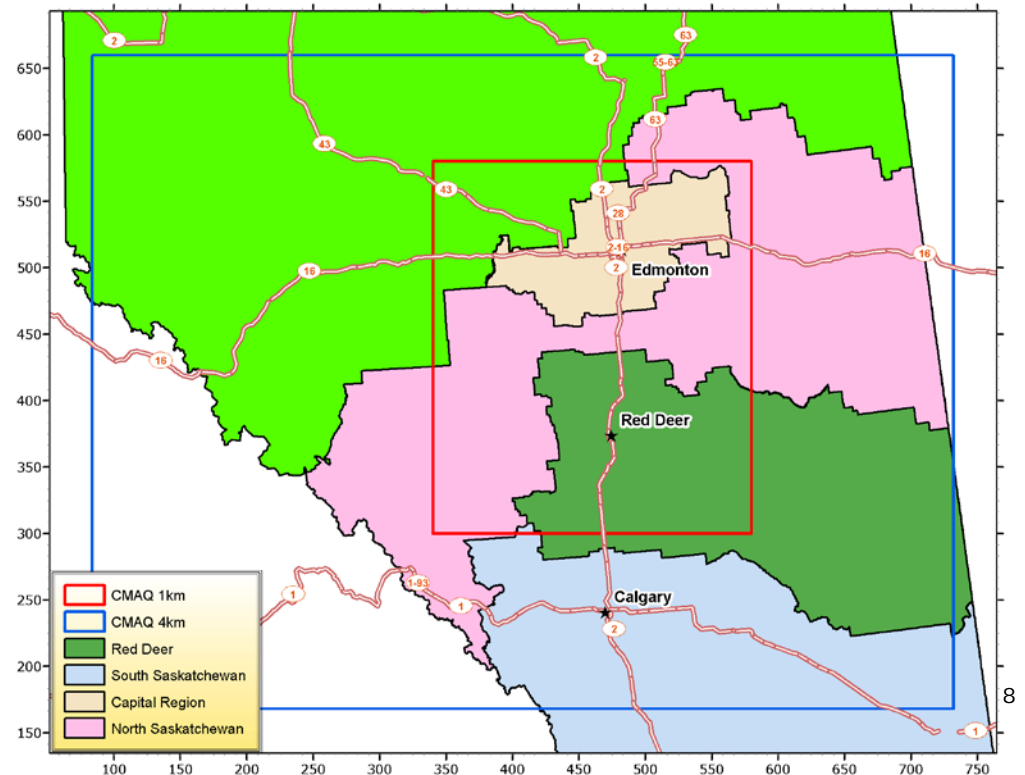
Colder temperatures favor NO<sub>3</sub> formation; less competition by SO<sub>4</sub> for NH<sub>3</sub>

## BACKGROUND: STUDY GOALS

- Develop a Photochemical Grid Model (PGM) modelling database for the Capital Region
  - Reproducing the observed winter elevated PM<sub>2.5</sub> sufficiently well
  - Tool for analyzing source contributions and control strategies
- Why PGM?
  - USEPA guidance (2014) requires use of photochemical models for secondary PM (e.g., sulphate and nitrate) and ozone.
  - Models with reduced form chemistry (e.g., CALPUFF) are incapable of properly simulating the chemical processes of secondary PM formation.

# MODELLING : OVERALL METHODOLOGY

- Inputs focused on Capital Region
  - 2010 Alberta inventory with extensive emissions inventory updates
    - Tremendous efforts in harmonizing multiple data sources, e.g., Environment Canada 2010, AEP's Industrial Survey, NPRI, other local inventories
    - Maximum allowance in EIAs not appropriate for this application
  - WRF meteorological modeling
- CMAQ setup
  - version 5.0.1
  - Resolution : 36/12/4/1.33 km & 22 layers
- Highest ranked PM episode
  - January 26 – February 4, 2010





# MODELLING: NITRATE FORMATION IN CMAQ

- NO<sub>x</sub> conversion to HNO<sub>3</sub> through OH and HO<sub>2</sub>



Winter: Scarce sunlight and radical availability

- Another gas-phase HNO<sub>3</sub> formation reaction is the reaction with aldehydes [formaldehyde (FORM) and acetaldehyde (ALD2)]:



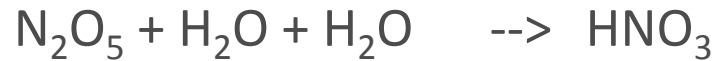
Primary sources of aldehydes in the Capital Region are from mobile sources and petrochemical plants

# MODELLING: NITRATE FORMATION IN CMAQ

- At night, HNO<sub>3</sub> formation occurs through N<sub>2</sub>O<sub>5</sub> (Dinitrogen Pentoxide),



NO<sub>3</sub><sup>-</sup> is formed through reactions involving NO<sub>2</sub> and ozone



Estimated water vapor by met. model becomes important.

- The final HNO<sub>3</sub> formation pathway in the CMAQ chemistry modules is the heterogeneous reaction probability (Y) of N<sub>2</sub>O<sub>5</sub>

Likely most important for Capital Region, but parameterization not designed for below 0° C (frozen particles)

- Availability of ammonia (NH<sub>3</sub>) to bind with gaseous HNO<sub>3</sub> to form particulate nitrate is important

Cooler and moister conditions favors NH<sub>4</sub>NO<sub>3</sub> over gaseous HNO<sub>3</sub>

# RESULTS: PHASE I INITIAL FINDINGS

- Except NO<sub>3</sub>, all species are over-estimated

Species	Average Obs	Average Model	Fractional Bias (%)	Fractional Error (%)
EC	1.5	5.6	127.6	127.6
NH4	4.6	10.7	72.8	76.3
NO3	10.3	4.3	-58.9	90.6
OC	3.1	9.0	121.6	121.6
SO4	5.3	25.8	129.8	129.8

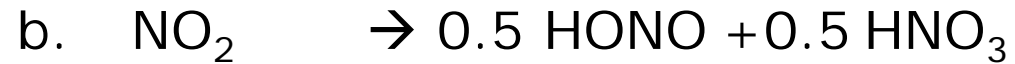
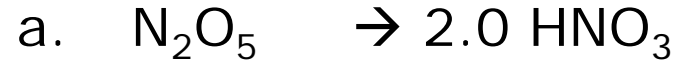
- Model performance evaluation
  - Individual PM species; compensating effects among species
  - Helps fine-tuning model through focused sensitivity cases

## RESULTS: SENSITIVITY TESTS AND PHASE II

- Key findings from 10 sensitivity tests (Phase I)
  - Under-estimation of nitrate is not  $\text{NH}_3$ -limited
  - Don't have enough  $\text{N}_2\text{O}_5$  to convert to  $\text{HNO}_3$
  - More radicals are needed to improve nitrate performance
  - Predicted  $\text{SO}_4$  is mostly secondary (conversion of  $\text{SO}_2$ ); suspect WRF moisture bias
- Phase II : revisit emissions inventory, meteorology and CMAQ assumptions
  - Improve emissions inventory (RWC, off-road)
  - WRF: remove moisture nudging which creates artificial cloud (7 tests)
  - CMAQ: heterogeneous pathway

# RESULTS: PHASE II FINDINGS

- HNO<sub>3</sub> pathway via heterogeneous reactions



$$k_{NO_2} = 3.0e-3 \text{ [1/min]} \times (S/V)$$

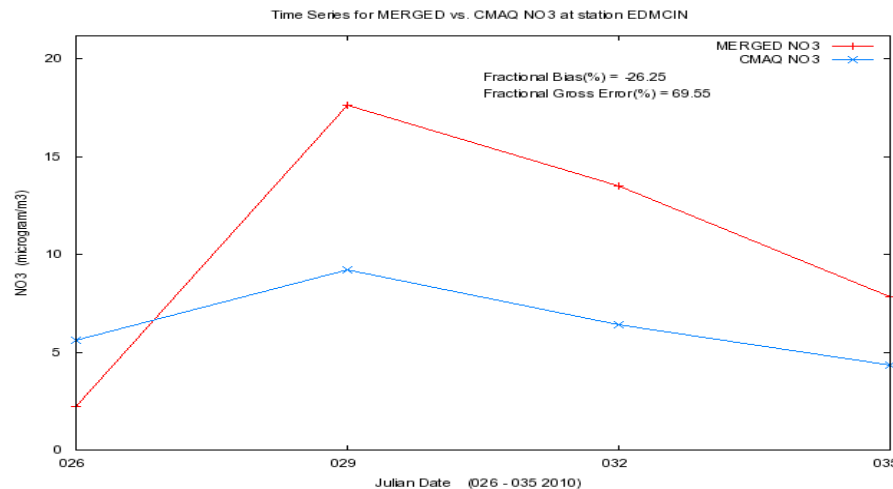
Kurtenbach et al. (2001)

Rxn a. is limited by N<sub>2</sub>O<sub>5</sub> availability (not enough ozone)

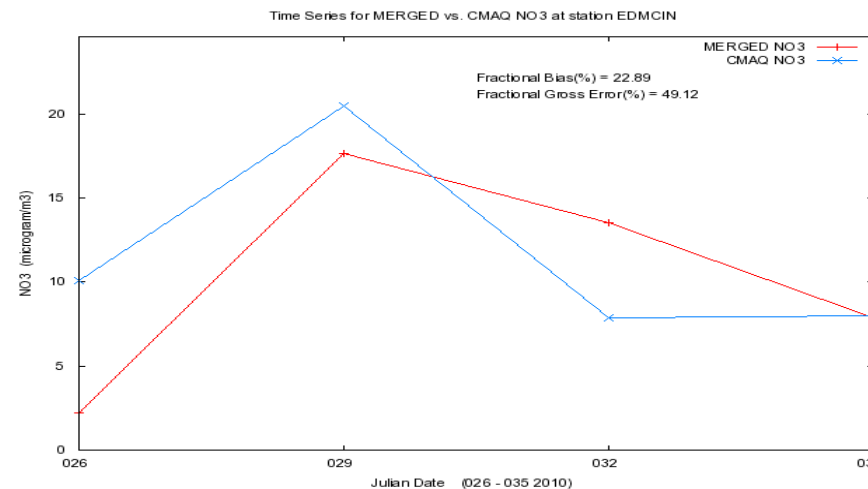
Rxn b. is important because of NO<sub>2</sub> is abundant

- Increasing kNO<sub>2</sub> by a factor of 100 increases NO<sub>3</sub> by a factor of two

### Default kNO<sub>2</sub>



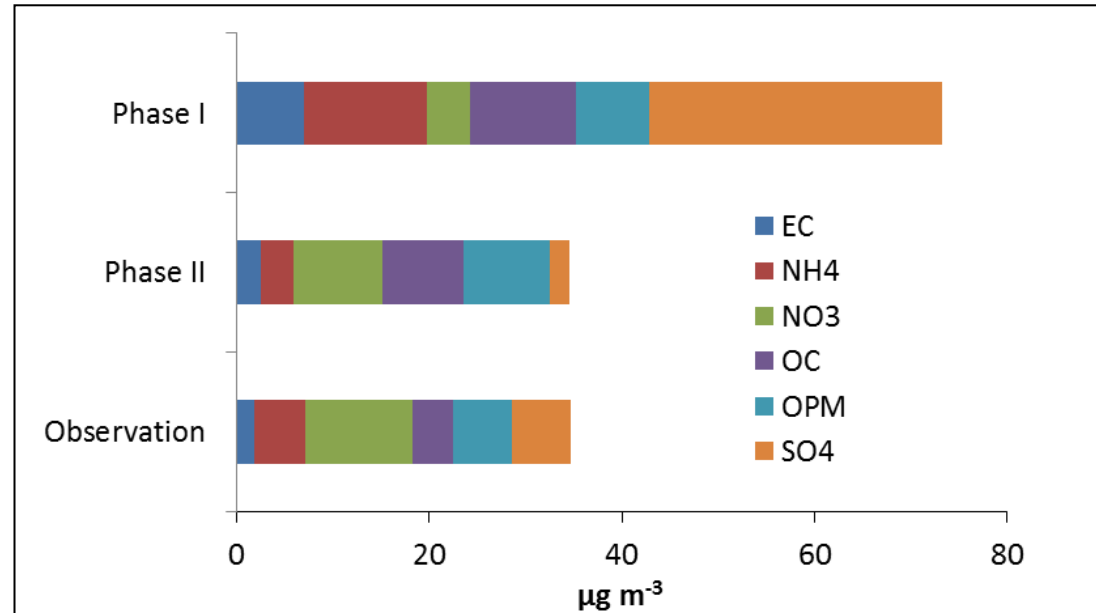
### kNO<sub>2</sub> x 100



Blue = Model  
Red = Obs.

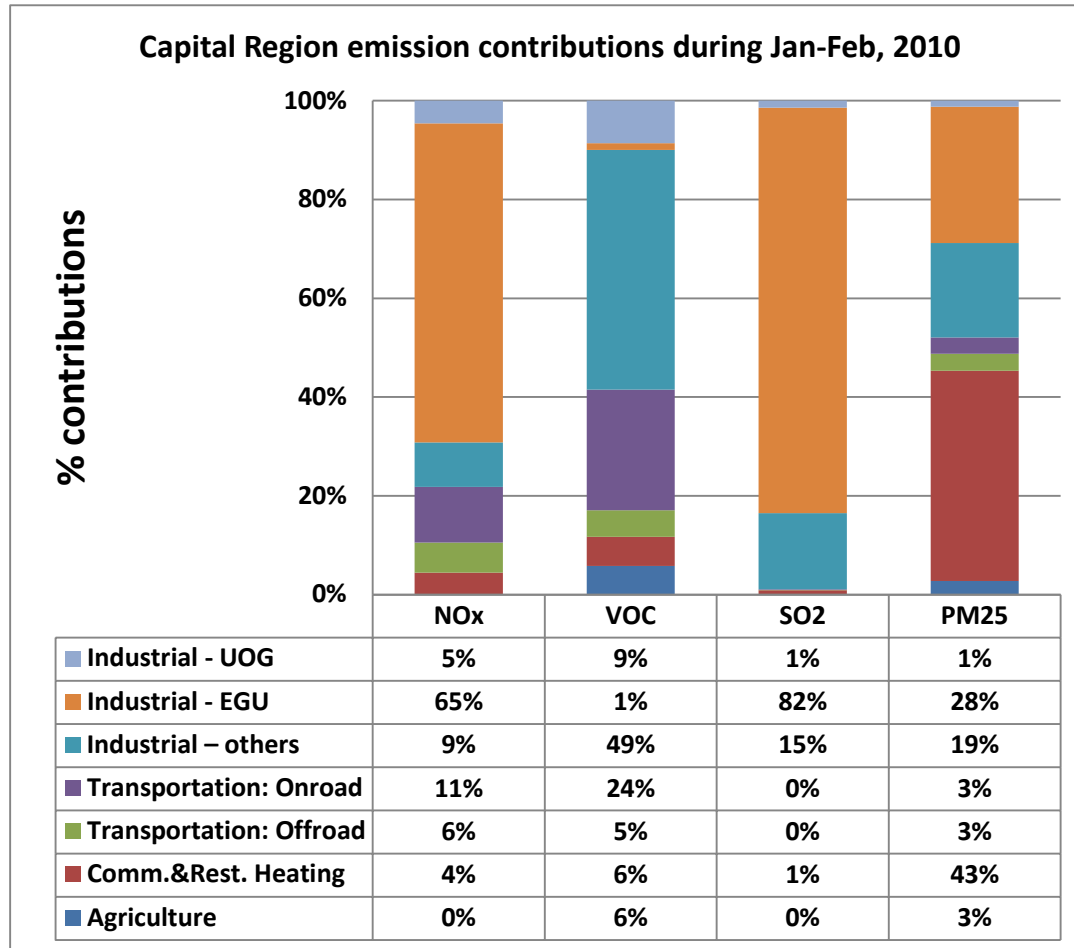
# RESULTS: PHASE II FINDINGS

- Significantly improved model performance in Phase II



- Abundant NO<sub>x</sub> in Edmonton limits ozone, hence limits reactions that rely on availability of radicals (such as  $\text{N}_2\text{O}_5 \rightarrow \text{HNO}_3$ )
- Need pathways that can go directly from NO<sub>x</sub> to HNO<sub>3</sub>
  - Too low  $k_{\text{NO}_2, \text{het}}$  for stagnant winter condition? Missing unknown pathway(s)?

# SOURCE ATTRIBUTION: CAPITAL REGION EMISSIONS (PHASE II)



Emissions attribution alone **cannot** tell a complete story

## Other factors

- Source location
- Stack parameters
- Composition of VOC emissions
- Meteorological conditions

## SOURCE ATTRIBUTION: MODELLING

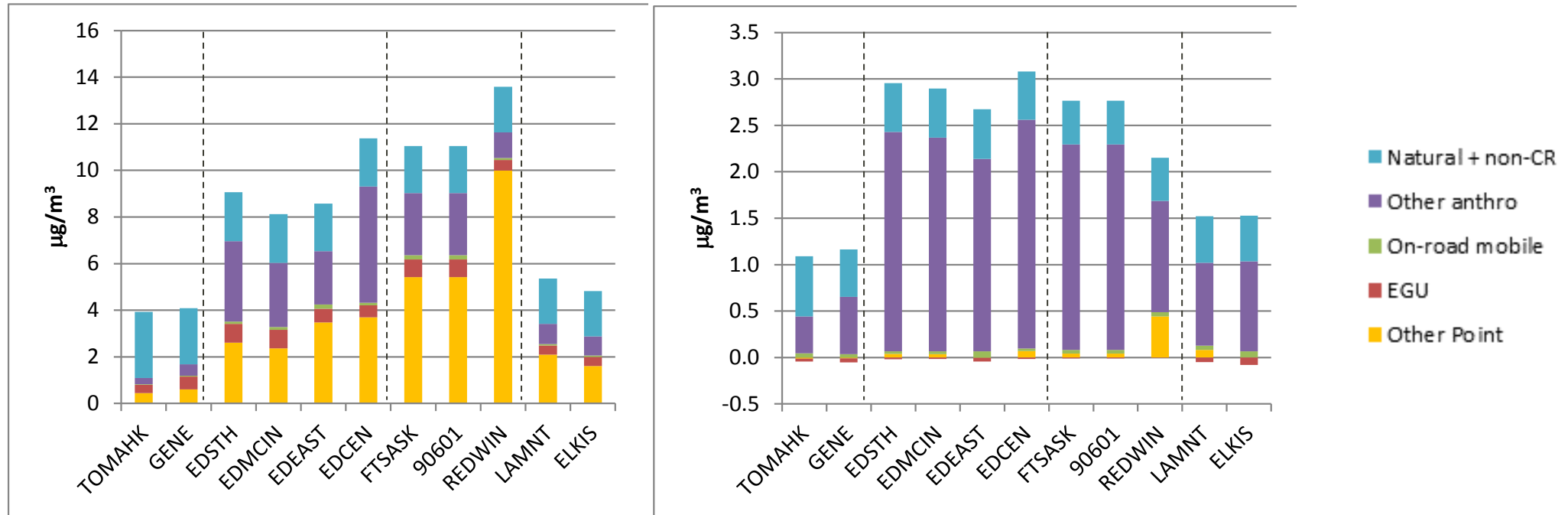
- Based on Phase I results (**subject to poor model performance**)
- Base Case setup with zero-out emissions to examine source contribution by sector
- Capital Region zero-out simulations:
  - On-road mobile (caveat: missing NH<sub>3</sub> emissions in Phase I)
  - Power Plant (EGU)
  - Other point sources - all stationary point sources except EGUs and UOG
  - All anthropogenic sources
- This approach can extend to quantify source contribution for each industrial source or sector (e.g., UOG, Refineries)



# SOURCE ATTRIBUTION: JAN-FEB AVERAGE

## SO4

## NO3



**Example** of contribution analysis at monitoring sites that can help identify major contributors

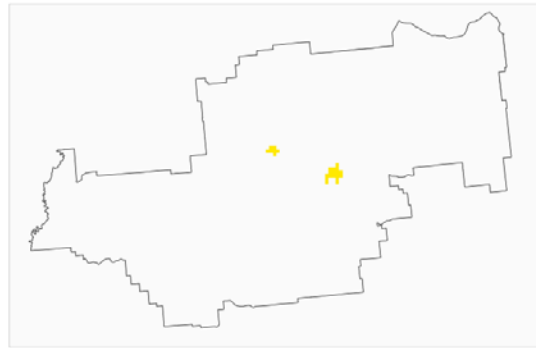
# SOURCE ATTRIBUTION: JAN-FEB AVERAGE

(a) On-road mobile

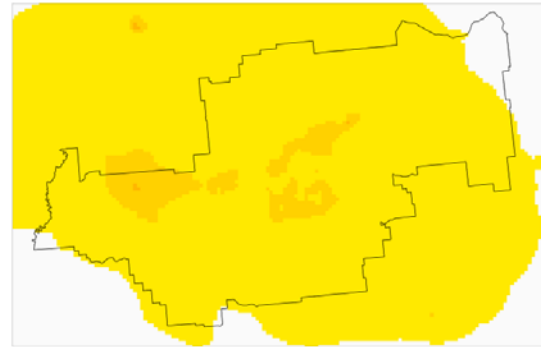
(b) EGU

(c) Other point sources

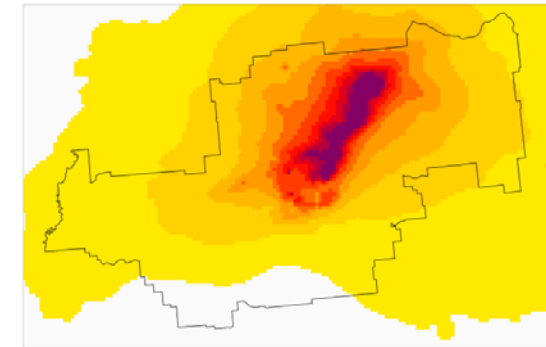
SO<sub>4</sub>



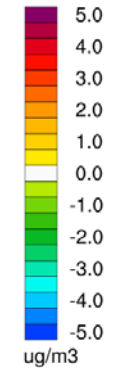
Min(1,99) = 0.00, Max(95,51) = 0.38



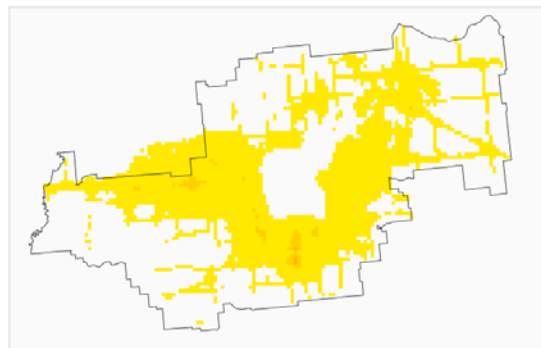
Min(16,1) = 0.04, Max(36,46) = 1.48



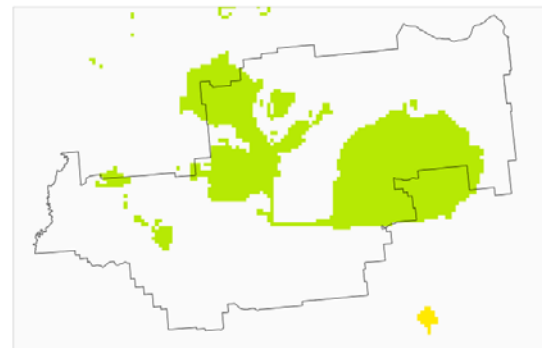
Min(46,1) = 0.02, Max(88,51) = 19.10



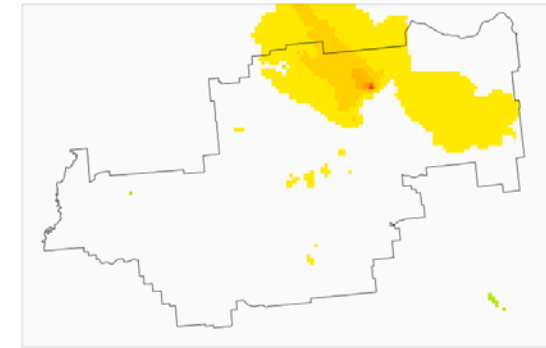
NO<sub>3</sub>



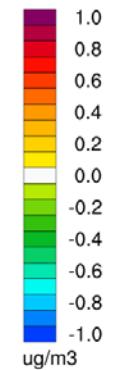
Min(144,98) = 0.00, Max(83,26) = 0.28



Min(60,71) = -0.12, Max(120,10) = 0.11



Min(76,81) = -0.11, Max(101,75) = 0.60



**Example** of contribution analysis that can help identify major contributors to the region

# CONCLUSION

- PGM is a powerful tool to help understand complex air quality issues
- Modeling inputs are important (emissions and meteorology)
- Model performance evaluation is necessary
  - PM chemistry is complex
  - limited by the observation data
- Source apportionment analysis is useful for developing effective emissions control programs
  - Single sources, source groups such as sectors, and sector/geographic area combination

# ACKNOWLEDGEMENT

- Funded by AEP
- Other staff at Ramboll Environ who participated in the study

## For more details...

Nopmongkol, U., Johnson, J., Brasher, B., Rasmussen, DJ, Shah, Tj., Zagunis, J., Liu., Z., Morris, R. 2015.

“Formation of Secondary PM<sub>2.5</sub> in the Capital Region Study.”

<https://open.alberta.ca/publications/formation-of-secondary-pm2-5>